

# Choosing the Right Grease Polymer for Water-Resistant Industrial Greases

Erik Willett, PhD – Functional Products Inc.

# Outline

- Our goal: teach fundamental material science of polymer selection
  - What is a “grease polymer”?
  - Interpenetrating Network Theory
  - Examples of use – water resistance
  - Key Points



## What makes a “grease polymer”?

# Polymers in Grease

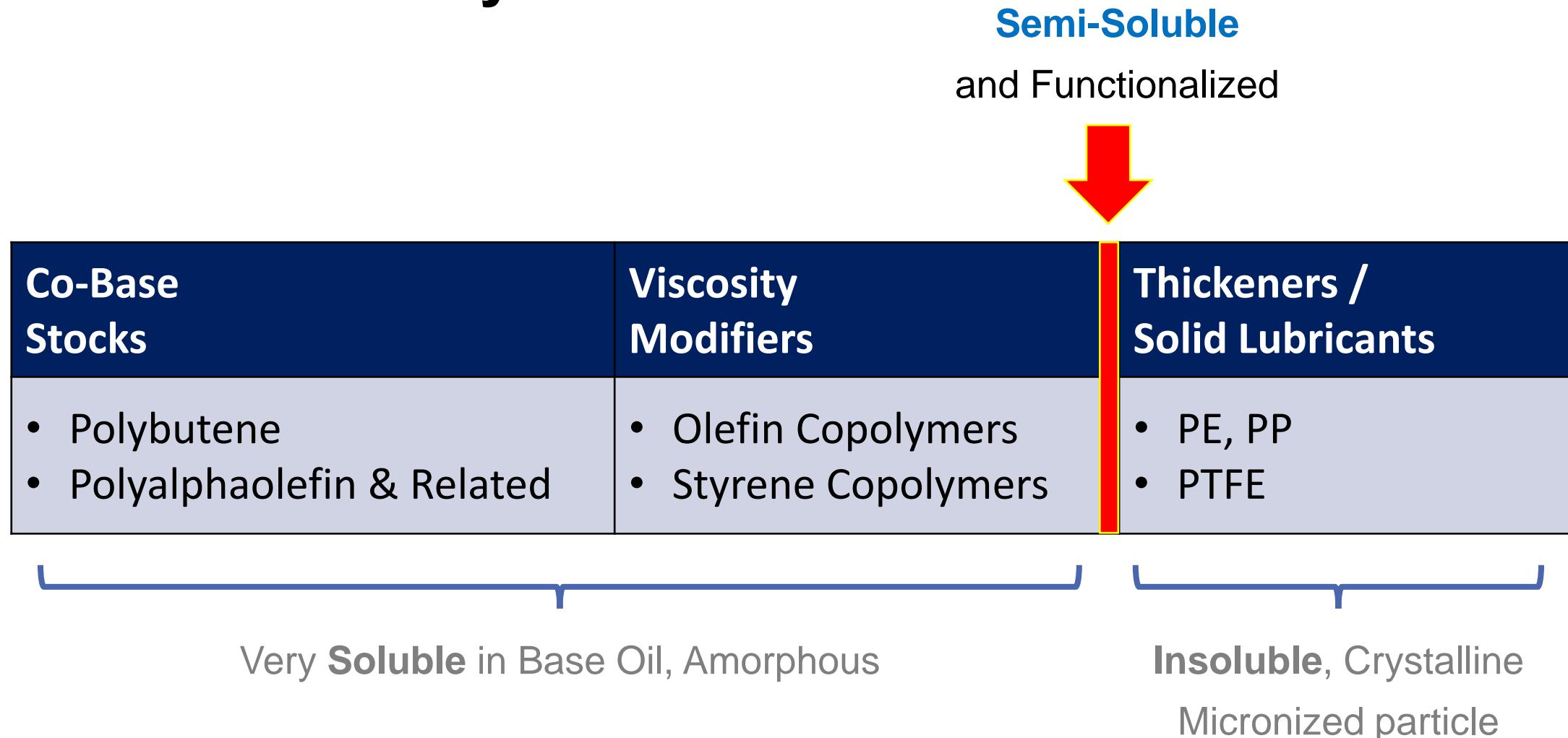
- Many polymers found in grease – soluble or insoluble

Co-Base Stocks	Viscosity Modifiers	Thickeners / Solid Lubricants
<ul style="list-style-type: none"><li>• Polybutene</li><li>• Polyalphaolefin &amp; Related</li></ul>	<ul style="list-style-type: none"><li>• Olefin Copolymers</li><li>• Styrene Copolymers</li></ul>	<ul style="list-style-type: none"><li>• PE, PP</li><li>• PTFE</li></ul>

Very **Soluble** in Base Oil, Amorphous

**Insoluble**, Crystalline  
Micronized particle

# But “Grease Polymers”...



# Grease Polymer Technology

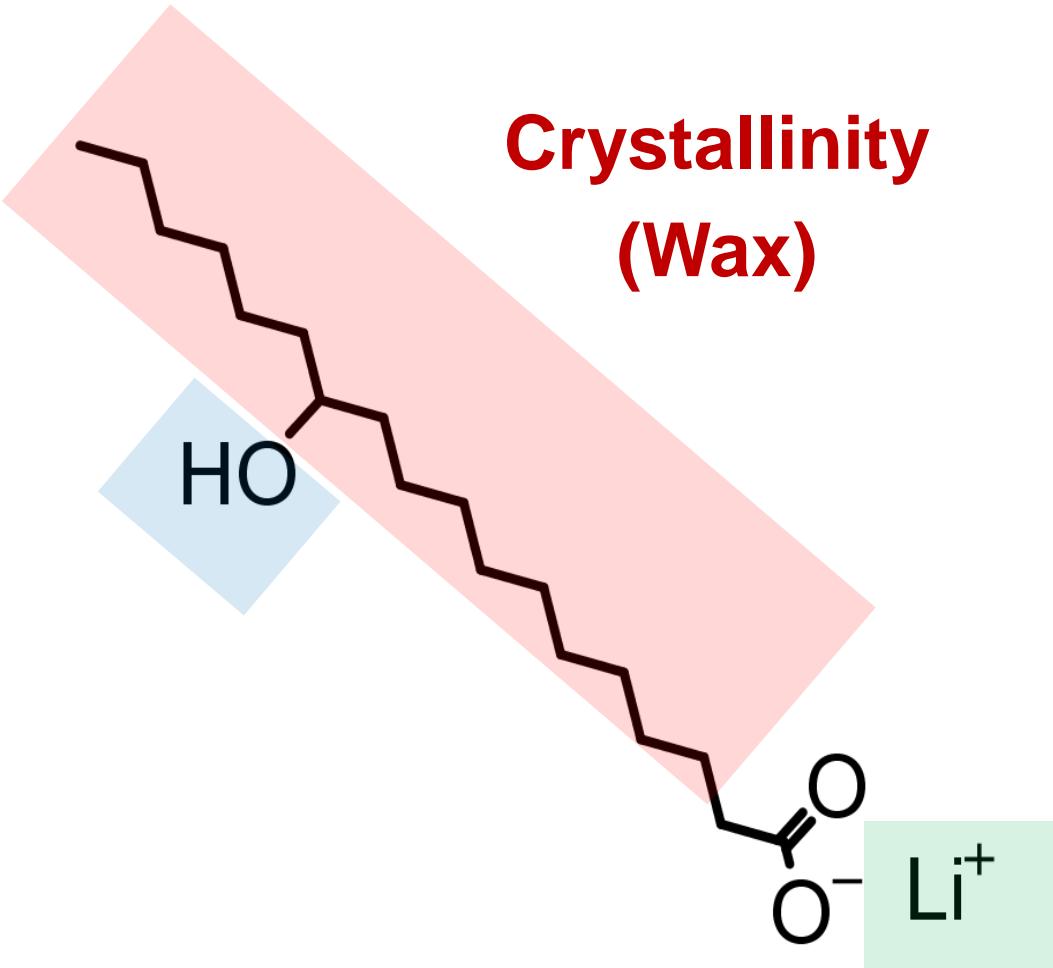
- We want a **composite** between thickener and polymer networks
- Three types of cohesive interactions to bind the networks together:
  1. Crystalline/waxy interactions
  2. Ionic interactions
  3. Hydrogen bonding

# Three Interactions

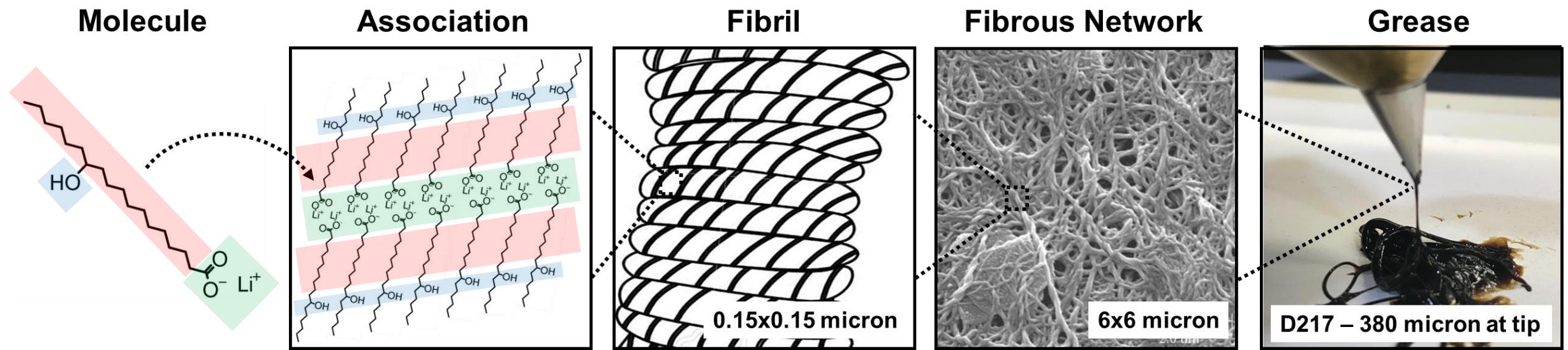
Hydrogen  
Bonding  
(-OH,-NH)

Crystallinity  
(Wax)

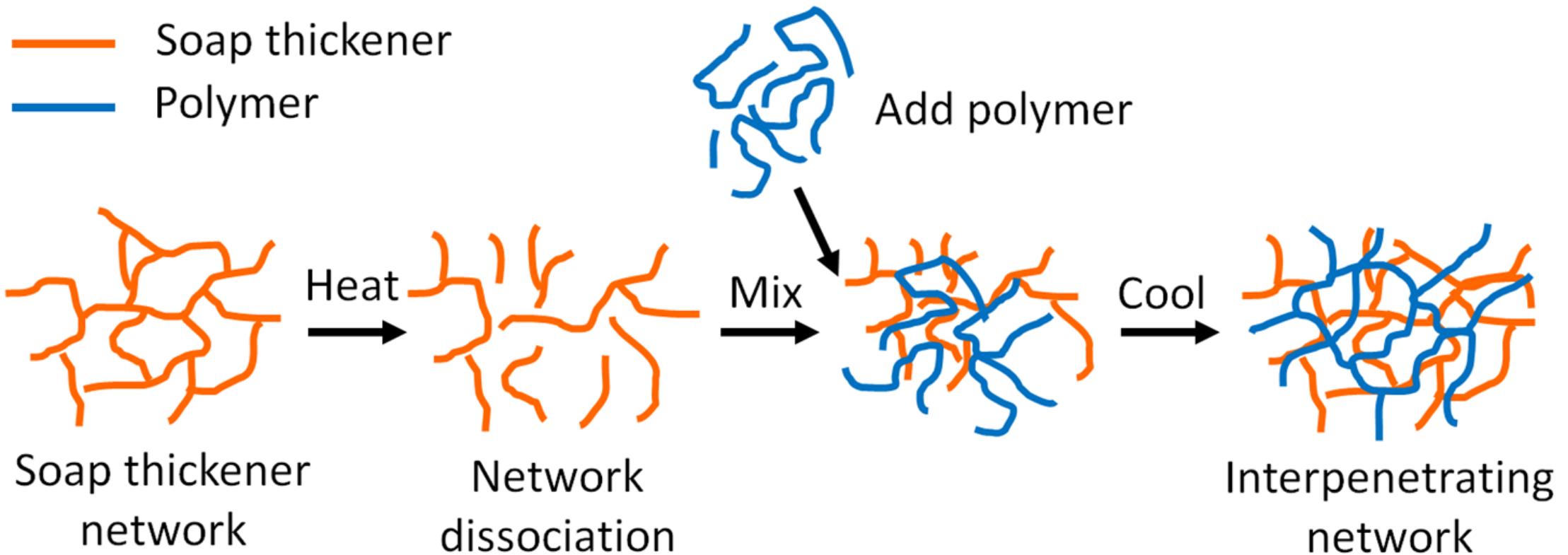
Ionic  
C+A-



# Self Assembly of Grease



# Interpenetrating Network (IPN) Structure



# Three Functionalities of Grease Polymers

1. Crystalline/Waxy



*think 'hook and loop' tape*

2. Reactive



*think super glue*

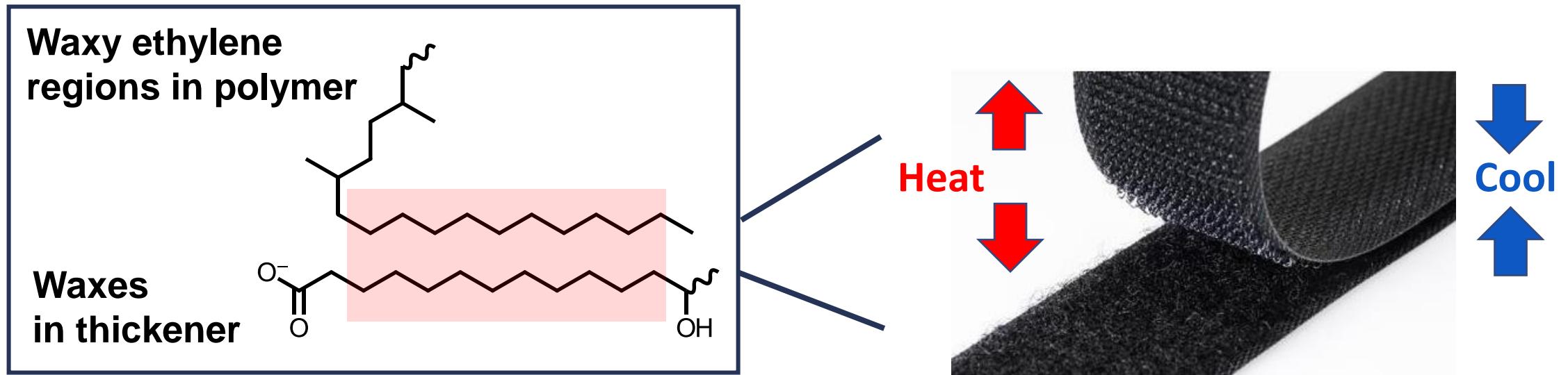
3. Hydrogen Bonding



*think polar attraction*

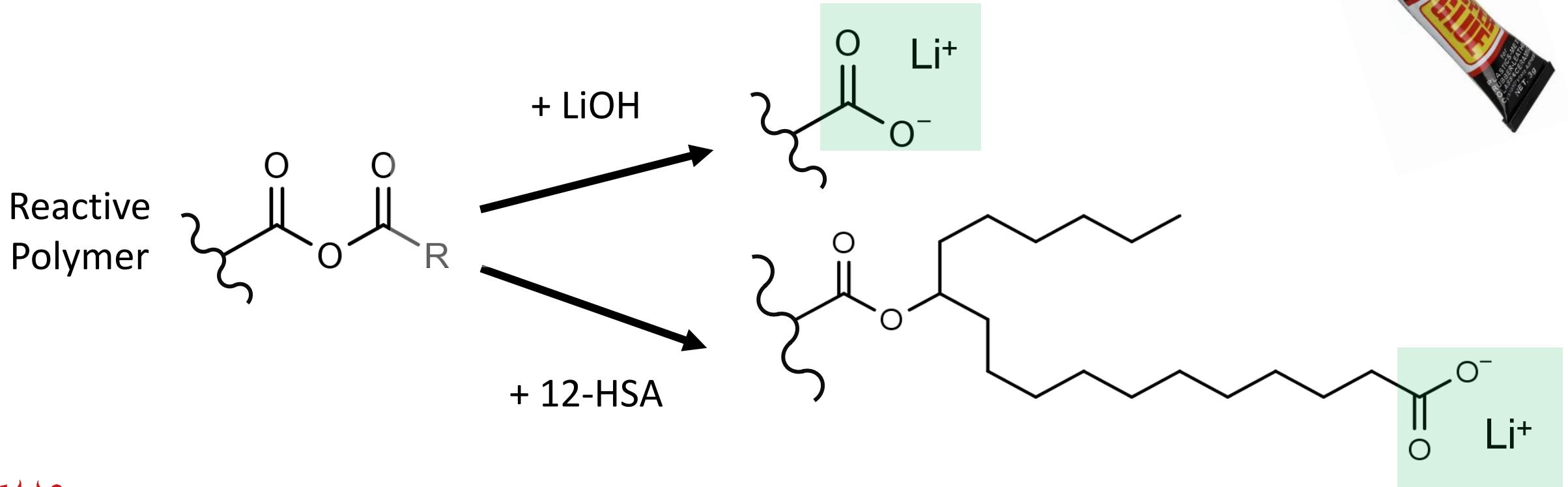
## 2 - Waxy Grease Polymers

- Lock in and interact with waxy segments of thickener
- Ethylene, alpha-olefins, styrene, etc.



## 2 - Reactive Grease Polymers

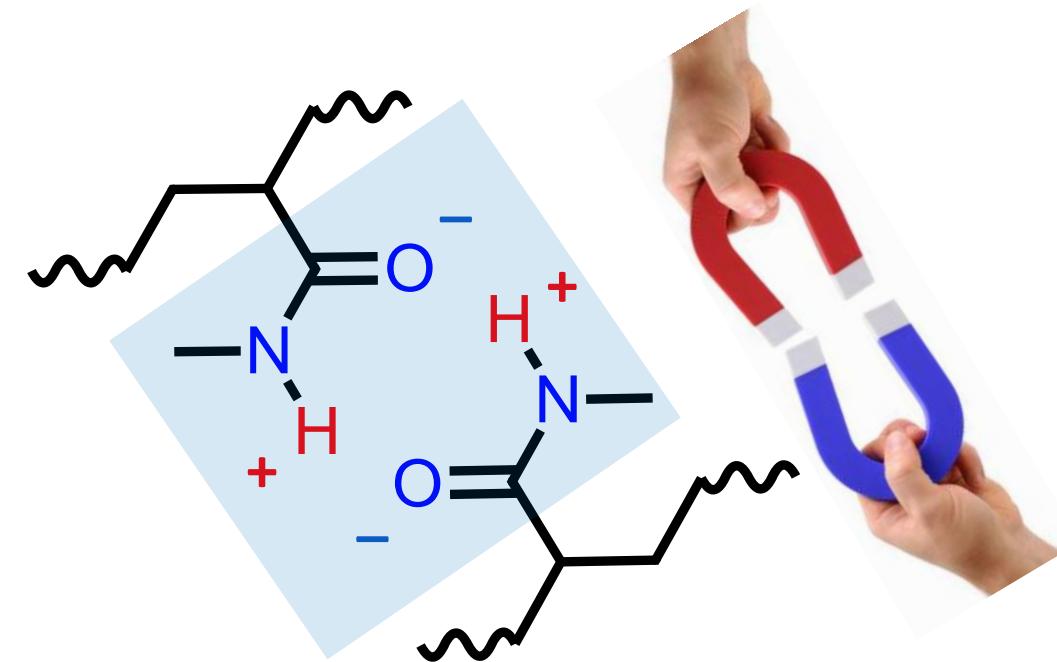
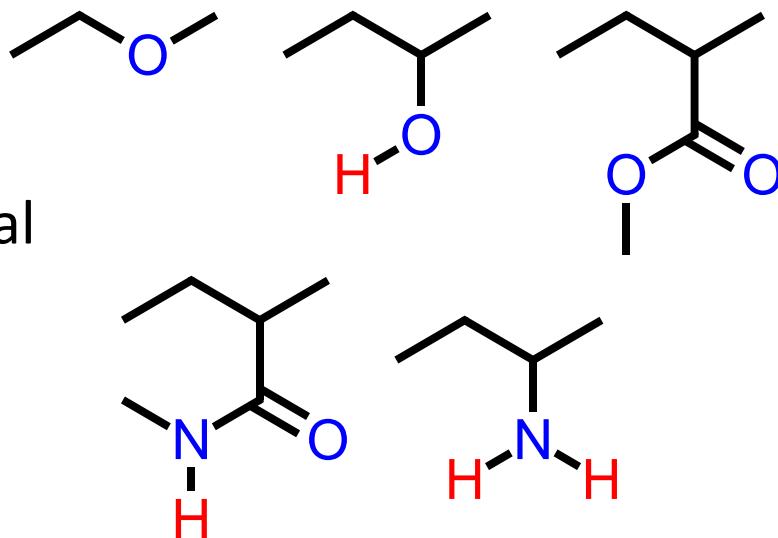
- Complex-like (diacid, borate ester) effect



# 3 - Hydrogen Bonding Grease Polymers

- Lewis acid/base electron donor or receptor
  - :O: or N:

CHON Functional Groups



# Thickener/Polymer Composite Effect

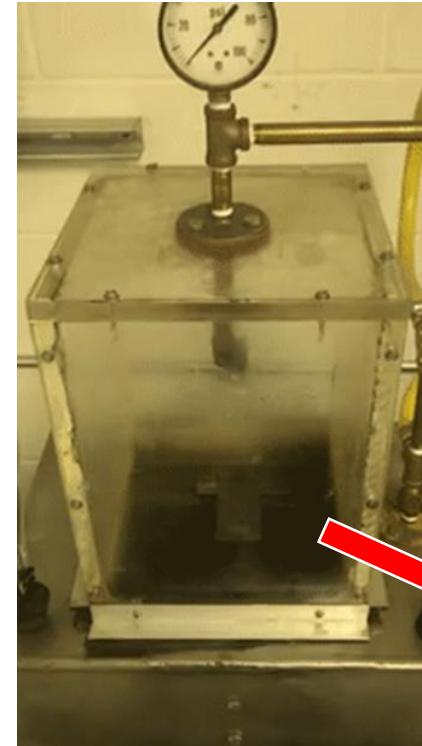
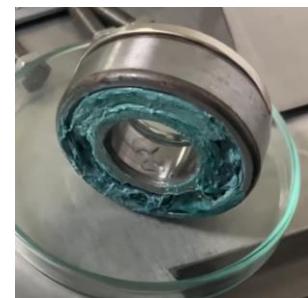
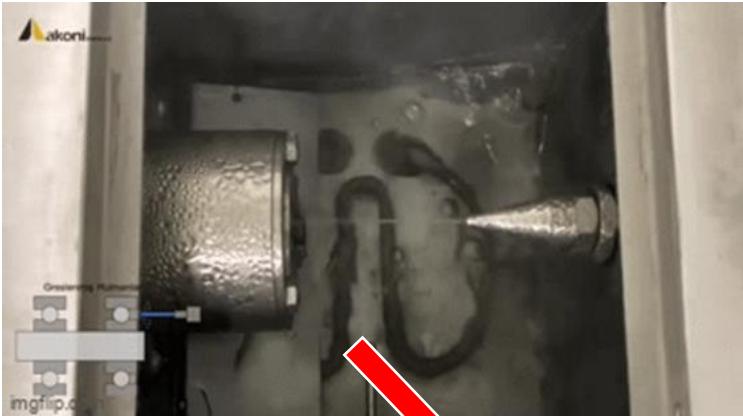
- Successful grease polymer selection meets two key criteria:
  1. Significant modification of physical properties...
    - Water resistance
    - Oil retention
    - Mechanical stability
    - Tackiness/adhesion
  2. ...at very low treat rates
    - 0.25 – 1.0wt%



## Examples of use – water resistance

# Water Resistance Testing

- Water Washout (D1264)
  - Favors adhesive grease
- Water Spray Off (D4049)
  - Favors cohesive grease

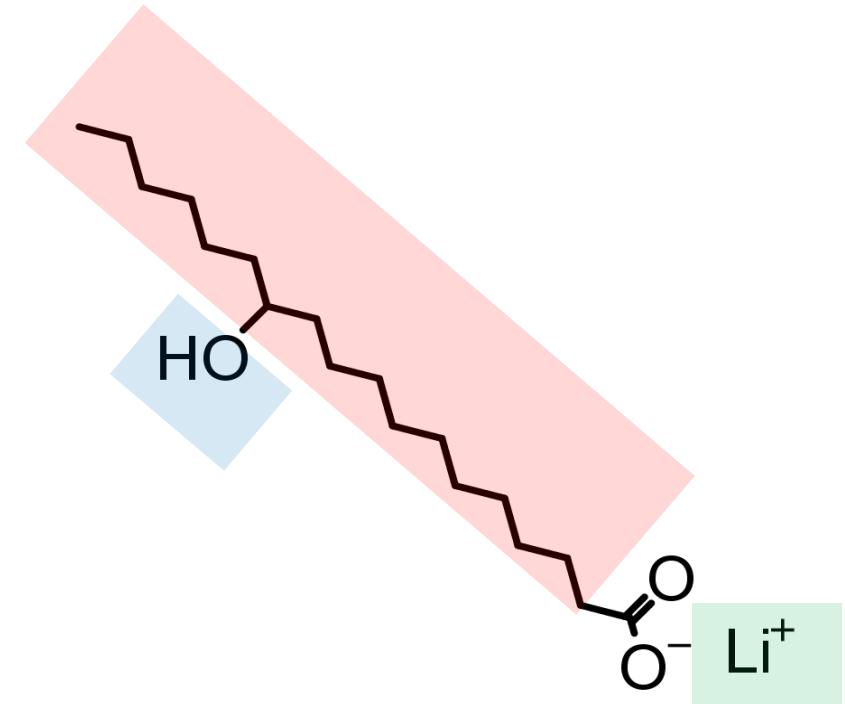


# Examples of WSO / WWO Effect

Treat	WSO (D4049)	WWO (D1264)
<b>NLGI #2 Lithium Complex (base)</b>	<b>52%</b>	<b>26%</b>
+0.25% reactive olefin copolymer pellet	26%	25%
+1% semi-crystalline copolymer flake	24%	11%
+1% styrene copolymer flake	9%	12%
+4% proprietary <i>blend</i> liquid	7%	2.0%
<b>NLGI #2 Calcium Sulfonate (base)</b>	<b>72%</b>	<b>4.8%</b>
+0.25% reactive olefin copolymer pellet	47%	3.9%
+0.5% reactive olefin copolymer pellet	4%	3.3%
+0.5% semi-crystalline copolymer flake	16%	4.2%
+1.0% styrene copolymer flake	7%	4.4%
<b>NLGI #2 Clay Grease (base)</b>	<b>49%</b>	<b>0.0%</b>
+5% concentrated olefin copolymer, liquid	48%	1.3%
+4% reactive olefin copolymer pellet, liquid	12%	-2%
+1% emulsion polymer	10%	29%

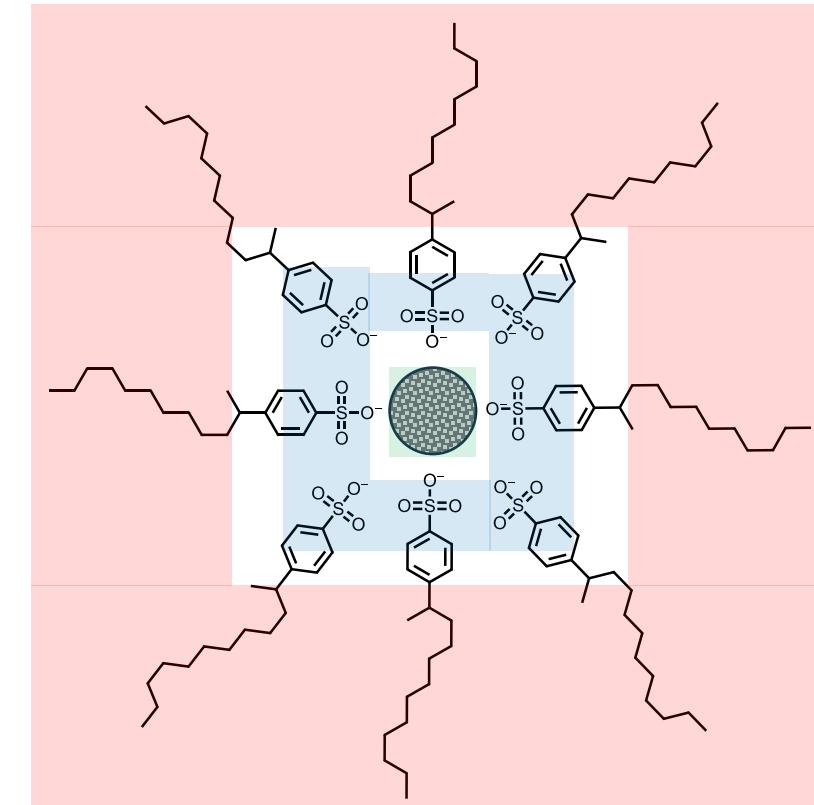
# Lithium and Lithium Complex

- Work best with reactive type polymer
  - Can add +0.5 or +1.0 NLGI grade
  - Less soap per kg of grease, save \$\$
- Waxy polymers tend to give bad texture



# Calcium Sulfonate (Complex)

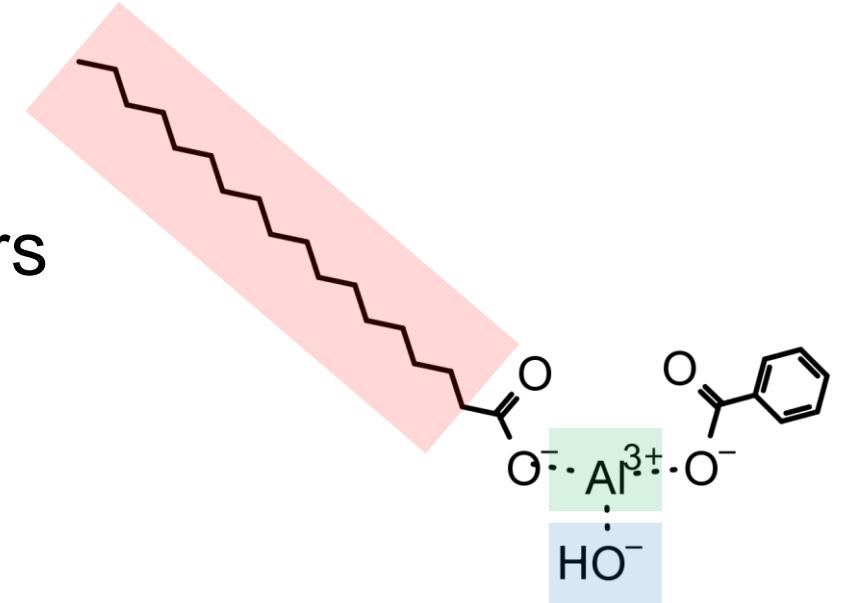
- Good inherent water washout resistance
- Poor **water sprayoff** resistance



- Work best with crystalline/waxy type polymer
  - Polar functionality can disrupt micelle structure

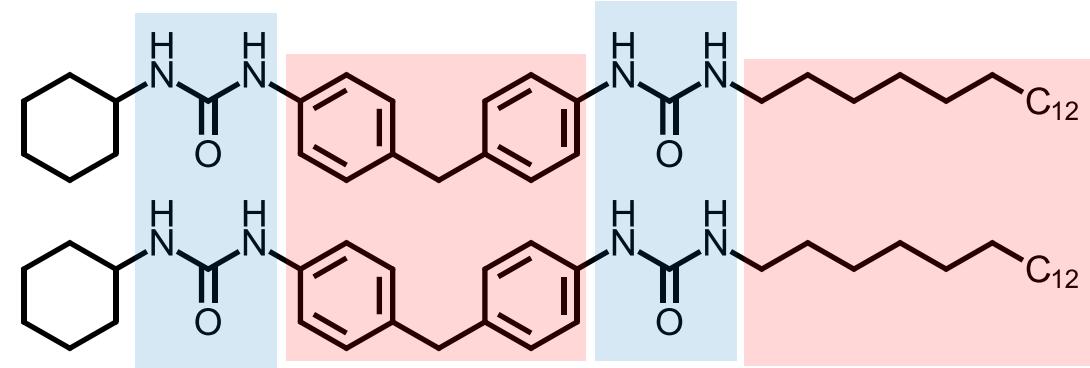
# Aluminum Complex

- Can have difficulty with water resistance
- Works well with waxy and reactive polymers
  - Reactive
    - Generally not HX-1
  - Waxy
    - HX-1 available



# Polyurea

- Strongly dominated by Hydrogen bonding Crystalline interactions (C18 + rings)
- Styrene copolymers have been most successful
- Hydrogen bonding can be antagonistic and disrupt thickener



# Difficulty with Full Synthetic Greases

- Polymer base oils are poor solvents for additives and very poor solvents for polymer additives
- Conventional polymers insoluble with high wt% mPAO or PB/PIB



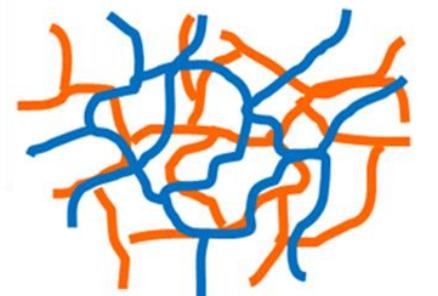
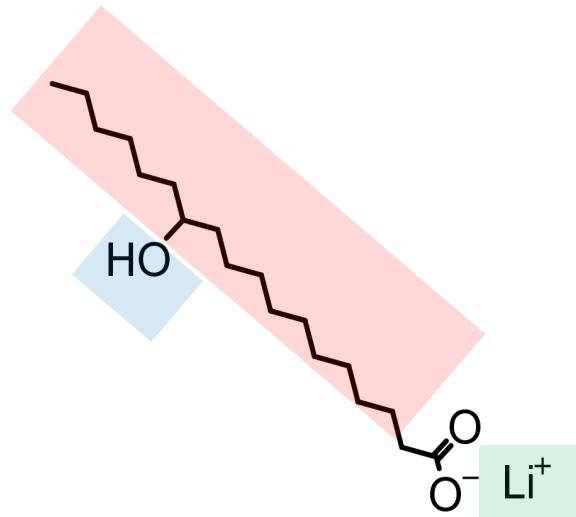
# Improper Selection Effects

- Disruption of grease thickener network
  - Loss of consistency (low yield)
- Dispersant-like effect
  - Increase in WSO or WWO %
- Excessive tackiness or a 'rubber-like' texture
  - Often occurs in naphthenic-based greases



## Conclusions

# Three Things to Remember



Interpenetrating  
network

1. Crystalline/Waxy



2. Reactive



3. Hydrogen Bonding



# More to Talk About in the Future

- How to add these grease polymers?
- What can grease polymers do for wet roll stability (D8022)?
- How does polymer affect low temperature, mechanical stability, and oil bleed?
- *Let's chat!*

# Key Points

- Grease polymers form composite with grease thickener
  - Small wt% treat, big effect
- The right polymer varies with thickener, base oil, viscosity
  - Polymer should have features similar to the interactions that bind the grease thickener network together

*Thank you for your attention*